## IN THE CLAIMS:

1. (Currently Amended) A solid electrolyte cell comprising:

a positive electrode having a positive electrode current collector and a positive electrode active material;

a negative electrode having a negative electrode current collector and a negative electrode active material; and

a solid electrolyte comprised of an electrolyte salt dispersed in a matrix polymer, said solid electrolyte being arranged between said positive electrode and the negative electrode 10; wherein,

a diene compound is contained in at least one of the positive electrode, negative electrode and the solid electrolyte, and

said diene compound is in a range of 0.0001 mol to 0.0005 mol per one gram of the positive electrode active material.

2. (Original) The solid electrolyte cell according to claim 1 wherein said diene compound is 1, 4-cyclohexadiene.

3. (Original) The solid electrolyte cell according to claim 1 wherein said diene compound is contained in said solid electrolyte.

## 4. (Cancelled)

5. (Currently Amended) A The solid electrolyte cell comprising:

according to claim 4 wherein said a solid electrolyte is made up of at least two layers,

namely a first solid electrolyte layer formed on the side positive electrode and a second solid

electrolyte layer formed on the side negative electrode; and

wherein the amount of said diene compound contained in said first solid electrolyte layer being not less than 75% of the total content thereof.

6. (Original) The solid electrolyte cell according to claim 1 wherein said solid electrolyte contains a non-aqueous solvent and is in a gelated state.

- 7. (Original) The solid electrolyte cell according to claim 1 wherein said matrix polymer is selected from the group consisting of polyethylene oxide, polypropylene oxide, polytetrafluoroethylene, polyvinylidene fluoride, polyvinylidene chloride, polymethacrylic acid, polyacrylic amide, polycarbonate, polysulfone and polyethersulfone.
- 8. (Original) The solid electrolyte cell according to claim 1 wherein said electrolyte salt is selected from the group consisting of LiPF6, LiClO4, LiCF3SO3, LiCF3SO3, LiAsF6, LiBF4, LiN(CF3SO3)2, C4F9SO3Li, LiC(CF3SO2)3, LiF, and LiBr.



- 9. (Currently Amended) The solid electrolyte cell according to claim 6 wherein said non-aqueous solvent is selected from the group consisting of ethylene carbonate, propylene carbonate, gamma a -butyrolactone, acetonitrile, diethylether, diethylene carbonate, dimethyl carbonate, 1, 2- dimethoxyethane, dimethyl sulfoxide, 1, 3-dioxolan, methyl sulfonate, 2-methyltetrahydrofuran, tetrahydrofuran, sulforan, 2, 4-difluoroanisole and vinylene carbonate.
- 10. (Original) The solid electrolyte cell according to claim 1 wherein said positive and negative electrodes are layered and coiled together with interposition of said electrolyte.
- 11. (Original) The solid electrolyte cell according to claim 1 wherein said positive and negative electrodes are layered and stacked together with interposition of said electrolyte.
- 12. (Original) The solid electrolyte cell according to claim 1 wherein a separator is arranged between said positive and negative electrodes.
- 13. (New Claim) A solid electrolyte cell according to claim 1, wherein a molar ratio of the diene compound to lithium of the positive electrode is from 0.01 to 0.05.